

Date \_\_\_\_\_ Name \_\_\_\_\_  
 Section \_\_\_\_\_ Team \_\_\_\_\_  
 Instructor \_\_\_\_\_

## Pre-Lab Study Questions

1. In making pickles, a cucumber is placed in a concentrated salt solution. Explain what happens to the cucumber when it is left in the salt solution for some time.

2. Why is it important that cell membranes are semipermeable?

3. What are the differences between solutions, colloids, and suspensions?

4. Indicate the compartment (A or B) that will increase in volume for each of the following pairs of solutions separated by a semipermeable membrane:

<b>Solution A</b>	<b>Solution B</b>	
a. 10% (m/v) starch	4% (m/v) starch	_____
b. 2% (m/v) albumin	5% (m/v) albumin	_____
c. 8% (m/v) sucrose	0.8% (m/v) sucrose	_____

5. Indicate whether a red blood cell will undergo hemolysis, crenation, or no change in each of the following:

- a. 10% (m/v) NaCl \_\_\_\_\_
- b. 2% (m/v) glucose \_\_\_\_\_
- c. H<sub>2</sub>O \_\_\_\_\_
- d. 5% (m/v) glucose \_\_\_\_\_

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## REPORT SHEET

# Properties of Solutions

### A. Identification Tests

Test	Observations for Positive Test
Cl <sup>-</sup> Test	
Starch Test	
Glucose Test	

### B. Osmosis and Dialysis

#### 1. Testing for Cl<sup>-</sup>, starch, and glucose

Time	0 min	30 min	Contents of Dialysis Bag
Cl <sup>-</sup> Test Observations			
Cl <sup>-</sup> present?			
Starch Test Observations			
Starch present?			
Glucose Test Observations			
Glucose present?			

#### Questions and Problems

Q1 Which substance(s) were found in the water *outside* the dialysis bag?

Q2 How did those substance(s) get into the water outside the dialysis bag?

**Q3** What substance(s) were retained inside the dialysis bag? Why were they retained?

**C. Filtration**

1. Appearance of filter paper \_\_\_\_\_
2. Substance present on filter paper \_\_\_\_\_

3. Identification Test	Observations	Substance Present
Cl <sup>-</sup> Test		
Starch Test		
Glucose Test		

4. Substance	Solution, Colloid, Suspension?	Reason
Cl <sup>-</sup>		
Starch		
Glucose		
Charcoal		

**Questions and Problems**

**Q4** State whether each of the following is isotonic, hypotonic, or hypertonic:

- a. H<sub>2</sub>O \_\_\_\_\_
- b. 0.9% (m/v) NaCl \_\_\_\_\_
- c. 10% (m/v) glucose \_\_\_\_\_
- d. 3% (m/v) NaCl \_\_\_\_\_
- e. 0.2% (m/v) NaCl \_\_\_\_\_

**Q5** Predict the effect on a red blood cell (crenation, hemolysis, or none) that the following solutions would have:

- a. 2% (m/v) NaCl \_\_\_\_\_
- b. H<sub>2</sub>O \_\_\_\_\_
- c. 5% (m/v) glucose \_\_\_\_\_
- d. 1% (m/v) glucose \_\_\_\_\_
- e. 10% (m/v) glucose \_\_\_\_\_